

Horsepower Calculations

Internal Combustion Chamber Pressure

<u>amount of HP to be developed</u>	<u>200</u>	<u>400</u>	<u>700</u>
<u>amount of kW</u> <u>1HP= .745 kW</u>	<u>149</u>	<u>298</u>	<u>521.5</u>
<u>amount of Btu/hr</u> <u>1kW= 3400 Btu/hr</u>	<u>506600</u>	<u>1013200</u>	<u>1773100</u>
<u>Efficiency</u>	<u>0.75</u>	<u>0.75</u>	<u>0.75</u>
<u>amount of H2 lb/hr needed</u> <u>H2 has 60,000Btu/lb</u> <u>considering efficiency</u>	<u>11.26</u>	<u>22.52</u>	<u>39.40</u>
<u>1 cylinder needs this lb/hr</u>	<u>2.25</u>	<u>4.50</u>	<u>7.88</u>
<u>in kg/hr</u>	<u>1.02</u>	<u>2.05</u>	<u>3.58</u>
<u>in g/min</u>	<u>17.06</u>	<u>34.11</u>	<u>59.70</u>
<u>need this g/rev</u> <u>2000 rpm motor</u>	<u>0.009</u>	<u>0.017</u>	<u>0.030</u>
<u>in g/explosion</u> <u>2rev= 1explosion</u>	<u>0.017</u>	<u>0.034</u>	<u>0.060</u>
<u>Designed Cylinder internal volume</u> <u>in Liters</u>	<u>1</u>	<u>1.5</u>	<u>2</u>
<u>Designed Cylinder explosion volume</u> <u>in Liters</u>	<u>0.1</u>	<u>0.15</u>	<u>0.2</u>
<u>internal pressure inside</u> <u>combustion chamber</u> <u>$P=nRT/V$</u> <u>n= number of moles in one explosion</u> <u>R=.082 liter-atm/mole-K</u> <u>T= 2273K (combustion temp.)</u> <u>V is explosion volume</u>	<u>15.90</u>	<u>21.19</u>	<u>27.82</u>

Fuel and compression needed for a 300 mile voyage

<u>No. of hours</u>	<u>4</u>	<u>4</u>	<u>4</u>
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<u>Total energy consumed (kWh)</u>	<u>596</u>	<u>1192</u>	<u>2086</u>
<u>Total Btu needed</u>	<u>2026400</u>	<u>4052800</u>	<u>7092400</u>
<u>Total Btu needed considering motor efficiency</u>	<u>2701867</u>	<u>5403733</u>	<u>9456533</u>
<u>No. of ft³ H2 needed H2 has 320 Btu/ft³ 1atm</u>	<u>8443</u>	<u>16887</u>	<u>29552</u>
<u>No. of liters H2 needed 1ft³=28.3 liters</u>	<u>239115</u>	<u>478230</u>	<u>836903</u>
<u>tank diameter</u>	<u>9</u>	<u>9</u>	<u>9</u>
<u>tank height</u>	<u>51</u>	<u>51</u>	<u>51</u>
<u>tank volume in liters</u>	<u>53</u>	<u>53</u>	<u>53</u>
<u>no. of tanks to be used</u>	<u>4</u>	<u>8</u>	<u>16</u>
<u>total available volume</u>	<u>213</u>	<u>425</u>	<u>850</u>
<u>compression needed in atm</u>	<u>1125</u>	<u>1125</u>	<u>984</u>
<u>compression needed in psi</u>	<u>15974</u>	<u>15974</u>	<u>13977</u>
<u>Expected H2 Recovery Rate in %</u>	<u>33%</u>	<u>33%</u>	<u>33%</u>
<u>compression needed given estimated recovery</u>	<u>10702</u>	<u>10702</u>	<u>9365</u>
<u>Quantum Technologies compression tanks</u>	<u>10000</u>	<u>10000</u>	<u>10000</u>
<u>Current Requirements</u>			
<u>H2 requirements g/min from line 21</u>	<u>17.1</u>	<u>34.1</u>	<u>59.7</u>
<u>H2O production g/min</u>	<u>154</u>	<u>307</u>	<u>537</u>
<u>Energy rate (J/min) needed to break this much water using 1g needs 10.5 J</u>	<u>1612</u>	<u>3224</u>	<u>5642</u>
<u>Energy rate in Watts (J/sec)</u>	<u>27</u>	<u>54</u>	<u>94</u>

<u>Voltage to operate electrolysis (V)</u>	<u>4</u>	<u>4</u>	<u>4</u>
<u>Current needed to operate electrolysis (amps)</u>	<u>6.7</u>	<u>13.4</u>	<u>23.5</u>
<u>Number of electrolysis units</u>	<u>4</u>	<u>6</u>	<u>8</u>
<u>current needed by each electrolysis unit</u>	<u>1.7</u>	<u>2.2</u>	<u>2.9</u>
<u>Expected battery needs (Wh) 10% of total energy</u>	<u>59600</u>	<u>119200</u>	<u>208600</u>
<u>Weight in kg at 100Wh/kg</u>	<u>596</u>	<u>1192</u>	<u>2086</u>